ABSTRACT

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Title:	CHARACTERIZATION OF THE MnO2-MWCNT NANOCOMPOSITE FOR SUPERCAPACITOR APPLICATIONS
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Year:	2020

Supercapacitors, with long cycle lives and long-term efficiency, have potential to become an appealing alternative to the lithium-ion battery; however, the specific energy of these devices lags well behind current state-of-the-art lithium-ion batteries. Nevertheless, numerous opportunities exist to optimize supercapacitors through the rational manipulation of the structure and morphology of novel materials. The material of interest for the current study is a composite of manganese dioxide (MnO2) and multi walled carbon nanotubes (MWCNT). Three concentrations of MWCNT, 5%, 16%, and 33% percent mass MWCNT relative to a constant concentration of MnO₂, were synthesized. Changes in nanostructure morphology of these samples were observed via transmission electron microscopy. A complex microstructure was observed that included a crystalline MnO2 adsorption onto the MWCNT wall surfaces (termed aggregation). MWCNT agglomeration was also noted as a dominant morphology. It is proposed that the aggregation ultimately gives rise to interfacial phenomena that effects MWCNT agglomeration. Aggregation width was studied as a function of concentration of MWCNT. Above 16% mass MWCNT, self-agglomeration of pristine carbon nanotubes is preferential over MnO2 aggregation. Preliminary results indicate that, as MWCNT concentration increases, there is a decline in average MnO₂ aggregation thickness. MnO₂ aggregation thickness may contribute directly to the specific capacitance of the composite material.